Questions: Use ESRT page 1 (Top, Left)

1. What is the half-life of Potassium-40?
2. What Isotope is used to date Human bones?
3. After $5.7 \times 10^3$ years, what percent of Carbon-14 remains undecayed?
4. After 1 half-life, what percent of Rubidium-87 remains undecayed?
5. What Isotope has the SHORTEST half-life?
6. What Isotope has the LONGEST half-life?

Fill out the data table below by calculating the % remaining after each half-life.

*** Only use the values from the row directly above the one you are working on ***

<table>
<thead>
<tr>
<th># Half-Lives</th>
<th>Number of Years (Age) 5,700</th>
<th>% Original C-14 remaining</th>
<th>Mass of C-14 remaining</th>
<th>Mass of N-14 (decay product)</th>
</tr>
</thead>
<tbody>
<tr>
<td>0</td>
<td>0 years</td>
<td>100%</td>
<td>10 grams</td>
<td>0 grams</td>
</tr>
<tr>
<td>1</td>
<td>5,700 yrs</td>
<td>50%</td>
<td>5 grams</td>
<td>5 grams</td>
</tr>
<tr>
<td>2</td>
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<td>3</td>
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<td>4</td>
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<td>5</td>
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<tr>
<td>6</td>
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<td></td>
</tr>
</tbody>
</table>

1. As $^{14}$C decays what happens to $^{14}$N?

Label which line represents $^{14}$C and which represents $^{14}$N.

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- **Represents**
- **-----** Represents
Relative Dating: The Order of Geologic Events

1. Relative dating puts geologic events or structures into proper chronological order. A rock’s relative age is its age compared to the ages of other rocks.

2. The principle of uniformitarianism states that the geologic processes that occurred in the past are basically the same as those that are occurring now.

3. The principle of superposition states that the bottom layer of a series of sedimentary layers is the oldest, unless the series has been overturned or thrust over by older rock.

4. Other clues that aid in determining a rock’s relative age include the following ideas:
   a. Rock layers are older than igneous intrusions that cut through them or igneous extrusions above them.
   b. Rocks are older than faults, joints, or folds that appear in them.
   c. Fragments of unmelted material occurring within a rock are older than the rock.
   d. In sedimentary rocks, the sediments are older than the cements that bind them and the rock formation itself.

Geologic Time Scale

1. Geologic history can be divided into time units (eras and periods) based on fossil evidence.

2. Most of the geologic past (Precambrian era) has left practically no fossil evidence.

3. Man’s existence is extremely short in comparison with the entire span of geologic time.

The Fossil Record

1. Fossil evidence suggests that a great variety of animals and plants have lived on Earth under a great variety of environmental conditions, and that most of them have become extinct.

2. Even though a large number of fossil types have been found, it is highly probable that a greater variety of organisms lived and died leaving no fossil trace.

3. Evolution is the gradual change in a species over long periods of time.

4. New studies of the fossil record have given rise to the theory of Punctuated Equilibria, which helps to explain the appearance of new species in relatively brief periods of geologic time.

Absolute Dating and Radioactive Decay

1. Used to give geologic events or structures an actual (absolute) age.

2. Radioactive decay is a process during which particles and/or electromagnetic energy are given off by atoms, and a new element (stable decay product) is formed.
   a. Decay of an individual atom occurs randomly
   b. The rate of radioactive decay occurs is predictable, and is a characteristic of an element. The rate is not affected by external factors, such as temperature, pressure, or chemical reaction.
   c. In a rock, as decay continues, the amount of the original radioisotope present decreases, and the amount of the stable decay product increases.

3. The length of time necessary for half a sample of a radioactive element to decay is the half-life of that element. Different radioisotopes have different half-lives.
   a. Radioisotopes with relatively short half-lives, such as C\(^{14}\), are used for dating recent organic remains. Isotopes with longer half-lives, such as U\(^{235}\) and K\(^{40}\), are used for dating older remains.
   b. By knowing the relative amounts of the original radioisotope and the decay product in a sample, and the half-life of the radioisotope, you can calculate the age of the sample.